

New Triterpenes from *Mangifera indica*

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NEW TRITERPENES FROM *MANGIFERA INDICA*MOHAMMAD ATAULLAH KHAN,* SHAIKH SIRAJUDDIN NIZAMI,
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ABSTRACT.—Two new cycloartane-type triterpenes have been isolated from the roots of *Mangifera indica*. Their structures were determined as cycloartan-3 β ,30-diol and cycloartan-30-ol, respectively, on the basis of chemical and spectroscopic methods.

Earlier investigations of *Mangifera indica* L. (Anacardiaceae) have led to the isolation of alkylgallates, amino acids, sugars, biflavones, and saponins (1-5). The present study on the roots of this species has yielded two new triterpenes, **1** and **2**.

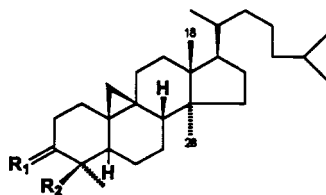
Compound **1** showed an $[M]^+$ peak at m/z 444.3928 (hrms), which corresponded to the molecular formula $C_{30}H_{52}O_2$ (calcd 444.3954) and indicated that **1** was a triterpene with five double-bond equivalents. The ir spectrum revealed the presence of an OH group (3460 and 3430 cm^{-1}), a cyclopropane ring (3040 cm^{-1}), and a *gem*-dimethyl group (1380 cm^{-1}). Additional bands between 1180 and 930 cm^{-1} agreed with a 3 β -OH, 5 α -structure (6).

The 1H -nmr spectrum ($CDCl_3$, 300 MHz) of **1** showed signals due to three secondary methyl groups (doublets at δ 0.90, $J=7.18$ Hz and δ 0.87, $J=6.72$ Hz), three tertiary methyl groups (singlets at δ 0.95, 0.92, and 0.88) and characteristic doublets at δ 0.56 ($J=4.18$ Hz) and δ 0.33 ($J=4.26$ Hz) for non-equivalent protons of a cyclopropyl me-

thylene group. In addition, there were signals due to a methine proton attached to a carbon bearing a hydroxyl group (δ 3.27, 1H, dd, $J_{a,a}=11.21$ Hz and $J_{a,c}=4.32$ Hz) and a methylene group of a primary alcohol with no proton on the adjacent carbon atom (δ 3.64, 2H, ABq, $J=12.78$ Hz). The ^{13}C -nmr spectrum showed 30 carbon atoms. The multiplicity of each carbon atom was determined using DEPT experiments (7,8) which revealed 6 methyl, 13 methylene, and 6 methine carbon atoms. The number of quaternary carbons was determined by subtracting these from the broad-band ^{13}C -nmr spectrum.

Both the 1H - and ^{13}C -nmr spectra indicated the presence of one primary and one secondary hydroxyl group in **1**, which were confirmed by acetylation of **1** to a diacetate **1a** and oxidation with Jones reagent to **1b** bearing both aldehydic and ketonic groups.

The mass spectrum of **1** was characteristic of cycloartane-type triterpenes (9,10). The genesis of various fragments was confirmed by link-scan measurements. The strain imposed on ring B was relieved by opening of the 9,10-bond followed by cleavage of the 5,6-bond and McLafferty rearrangement. The characteristic ion at m/z 288.2798 ($C_{21}H_{36}$) represented the fragmentation induced by a cyclopropane ring in 9,19-cyclosterols and related tetracyclic triterpenes (11). Loss of the side-chain from this ion gave another fragment at m/z 175.14161 ($C_{13}H_{19}$). On the other hand, the direct loss of the side-chain from the parent ion



- 1** $R_1=\alpha H, \beta OH$ $R_2=CH_2OH$
1a $R_1=\alpha H, \beta OAc$ $R_2=CH_2OAc$
1b $R_1=O$ $R_2=CHO$

gave a fragment at m/z 331.2601 ($C_{22}H_{35}O_2$). Further ions at m/z 429.3689 ($C_{29}H_{49}O_2$), 426.3837 ($C_{30}H_{50}O$), and 411.3597 ($C_{29}H_{47}O$) originated by the loss of a methyl radical, H_2O and methyl plus H_2O , respectively, from the molecular ion. It was observed that all of these fragment ions were formed through the same route in cycloartanol (9,10), which helped to show that compound **1** differs from cycloartanol only in having one additional hydroxyl group.

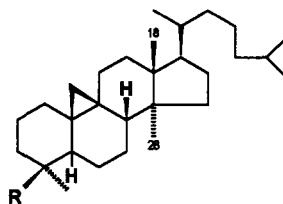
The ions at m/z 331.2601 and 175.1461 suggested that both hydroxyl groups are present in rings A and B. This conclusion was further supported by nmr chemical shifts of various carbon atoms of rings C and D and the side-chain, which were in close agreement with those of cycloartanol.

On biogenetic grounds and by 1H - 1H -correlated nmr spectroscopy, the secondary hydroxyl group was assigned to C-3. The carbinyl proton at δ 3.27 showed cross-peaks with two other protons limiting it to either position 1 or 3. However, the chemical shifts of C-1 through C-3 showed close correlations with those of cycloartanol providing evidence for the hydroxyl group being affixed to C-3 rather than C-1. The chemical shift and coupling constants of the carbinyl proton are in accord with its axial and α -orientation.

The remaining question was the position of the primary hydroxyl group. Careful comparison of the ms and ^{13}C -nmr spectrum of **1** with those of cycloartanol indicated that the primary hydroxyl group could be assigned to either C-29 or C-30. The latter proved to be correct by interpretation of 1H -nmr data for the methylene group in **1** and **1a** which agreed with those for axially oriented- CH_2OH and $-CH_2OAc$ groups, respectively (12). The 1H -nmr spectrum of **1b** also showed a signal characteristic of an aldehydic proton at δ 9.71, corresponding to an axial orientation (13). The long-range 1H - ^{13}C -correlated spectrum

(COLOC) of **1b** showed a cross-peak between the aldehydic carbon and the methyl protons of C-29. Hence the hydroxymethyl group in **1** was assigned to C-30 in a β - and axial orientation. The stereostructure of this compound is therefore represented as **1**.

Compound **2** showed an $[M]^+$ peak at m/z 428.3977 (hrms) corresponding to a molecular formula of $C_{30}H_{52}O$ (calcd 428.4005), indicating five double-bond equivalents in the molecule. The ir spectrum showed an OH group (3420 cm^{-1}), a cyclopropane ring (3040 cm^{-1}), and a *gem*-dimethyl group (1375 cm^{-1}).



- 2** R=CH₂OH
2a R=CH₂OAc
2b R=CHO

The 1H -nmr spectrum of **2** showed signals for three secondary methyl groups (doublets at δ 0.90, $J=6.98$ Hz and δ 0.86, $J=6.72$ Hz) three tertiary methyl groups (singlets at δ 0.94, 0.91, and 0.88) besides the characteristic doublets at δ 0.56 ($J=4.21$ Hz) and δ 0.33 ($J=4.28$ Hz) for a cyclopropane ring. In addition, there was a signal due to a methylene group for a primary alcohol with no proton on the adjacent carbon atom (δ 3.62, 2H, ABq, $J=12.18$ Hz). The ^{13}C -nmr spectrum showed 30 carbon atoms whose multiplicities were determined by DEPT experiments (7,8), which revealed the presence of 6 methyl, 14 methylene, and 5 methine carbon atoms. Both the 1H - and ^{13}C -nmr spectra showed one primary hydroxyl group in **2** which was confirmed by acetylation of **2** to a monoacetate **2a** and oxidation with Jones reagent to **2b**.

The mass spectrum of **2** was also characteristic of cycloartane-type triterpenes (9,10). The diagnostic ions at m/z 288.2796 ($C_{21}H_{36}$) and 175.1462 ($C_{13}H_{19}$) indicated that compound **2** is a tetracyclic triterpene (11). Further ions at m/z 315.2651 ($C_{22}H_{35}O$), 413.3738 ($C_{29}H_{49}O$), 410.3886 ($C_{30}H_{50}$) and 395.3645 ($C_{29}H_{47}$) were generated by the loss of side-chain, methyl radical, H_2O and methyl plus H_2O , respectively, from the molecular ion. All of these fragments have compositions similar to those observed for the corresponding ions of cycloartanol, demonstrating that compound **2** is an isomer of cycloartanol, the former differing from the latter in having a primary alcoholic group in ring A. This was further supported by chemical shifts of various carbon atoms of rings B, C, D, and the side-chain, which showed close agreement with those of cycloartanol.

On the basis of ms and ^{13}C -nmr data it was concluded that the hydroxymethyl group in **2** is at C-29 or C-30. The latter was proved to be correct by analysis of the 1H -nmr data of **2** and **2a**, which agreed with the presence of an axial $-CH_2OH$ and $-CH_2OAc$ group, respectively (12). The aldehydic proton at δ 9.72 in **2b** also confirmed axial orientation (13). The COLOC experiment revealed a cross-peak between carbon and methyl protons of C-29. The hydroxymethyl group in **2** was therefore assigned to C-30 in a β -axial orientation and the stereostructure of compound **2** is as shown in **2**.

EXPERIMENTAL

GENERAL EXPERIMENTAL PROCEDURES.—The ir spectra were recorded in $CHCl_3$. DEPT nmr experiments were carried out with $\theta=45^\circ$, 90° , and 135° , with quaternary carbons being determined by subtraction of these spectra from the respective broad-band ^{13}C -nmr spectra. The hrms were recorded on a double-focusing instrument coupled to a PDP 11/34 computer system. The 1H -nmr spectra ($CDCl_3$) were recorded at 300 MHz with TMS as internal reference. The 2D COSY-45 experiments were acquired at 300 MHz with a sweep width of 4000 Hz (2K data points in w_2) and 2000 Hz (256 t_1 values zero-filled to 1K) in w_1 . The heteronuclear two-dimensional 1H - ^{13}C

chemical shift correlation experiments were carried out at 300 MHz with a sweep width of 12820 Hz (2K data point in w_2) and 1024 Hz (256 t_1 values zero-filled to 2K) in w_1 . In both the 2D nmr experiments a 2 sec relaxation delay was used and 16 transients were performed for each t_1 value.

PLANT MATERIAL.—*Mangifera indica* roots were collected from the Karachi region and were identified by the Department of Botany, University of Karachi. A voucher specimen has been deposited in the herbarium of the Department of Botany, University of Karachi (voucher No. KUH 4378).

EXTRACTION AND ISOLATION.—The air-dried roots (2 kg) were crushed into small pieces and then extracted three times with MeOH (total 5 liters). The combined MeOH extract was evaporated under reduced pressure to afford a gummy residue which was partitioned between hexane and H_2O . The residue recovered from the hexane fraction was chromatographed over activated Si gel and elution was carried out with solvent gradients of increasing polarity consisting of hexane, hexane/ $CHCl_3$, $CHCl_3$, $CHCl_3/MeOH$, and MeOH. The hexane- $CHCl_3$ eluents (4.5:5.5 and 4:6) yielded a crystalline residue which, on repeated crystallization from a mixture of $CHCl_3$ and MeOH, provided colorless fine needles of compound **2** (16 mg) and compound **1** (18 mg), respectively.

Cycloartan-3 β ,30-diol [**1**].—Mp 196–198°; $[\alpha]_D +34.9^\circ$ ($c=0.18$, $CHCl_3$); ir, see text; 1H nmr ($CDCl_3$, 300 MHz) δ 3.64 (2H, ABq, $J=12.78$ Hz, H₂-30), 3.27 (1H, dd, $J_{ax}=11.21$ Hz, $J_{ax'e}=4.32$ Hz, H-3), 0.95 (3H, s, Me-18), 0.92 (3H, s, Me-29), 0.90 (6H, d, $J=7.18$ Hz, Me-26 and -27), 0.88 (3H, s, Me-28), 0.87 (3H, d, $J=6.72$ Hz, Me-21), 0.56 and 0.33 (2H, dd, $J=4.18$ and 4.26 Hz, H₂-19); ^{13}C nmr ($CDCl_3$, 75.43 MHz) δ 31.91 (C-1), 30.33 (C-2), 78.60 (C-3), 39.61 (C-4), 47.08 (C-5), 21.00 (C-6), 28.08 (C-7), 47.84 (C-8), 20.10 (C-9), 26.20 (C-10), 26.01 (C-11), 35.62 (C-12), 45.13 (C-13), 48.79 (C-14), 32.71 (C-15), 26.51 (C-16), 52.21 (C-17), 17.91 (C-18), 29.81 (C-19), 36.01 (C-20), 18.30 (C-21), 36.42 (C-22), 24.01 (C-23), 39.41 (C-24), 28.21 (C-25), 22.51 (C-26), 22.71 (C-27), 19.32 (C-28), 22.01 (C-29), and 63.14 (C-30); ms m/z 444, 429, 426, 411, 331, 288, 175.

ACETYLATION OF 1.—Compound **1** (5 mg) was refluxed with Ac_2O (2.5 ml) in pyridine (1 ml) for 30 min. The reaction mixture was worked up in the usual manner to yield acetate **1a** (3.7 mg) that was crystallized from $CHCl_3/MeOH$, mp 212–214°; $[\alpha]_D +31.2^\circ$ ($c=0.12$, $CHCl_3$); ir ν_{max} 3040, 1720, 1380, and 1220 cm^{-1} ; 1H nmr ($CDCl_3$, 300 MHz) δ 3.94 (2H, ABq, $J=12.76$ Hz, H-30), 3.91 (1H, dd, $J_{ax}=11.28$ Hz, and $J_{ax'e}=4.18$ Hz, H-3), 2.13 (3H, s, OAc), 2.10 (3H, s, OAc), 0.95 (3H, s, Me-18), 0.93 (3H, s, Me-29),

0.90 (6H, d, $J=7.12$ Hz, Me-26 and -27), 0.88 (3H, s, Me-28), 0.87 (3H, d, $J=6.81$ Hz, Me-21), 0.57 and 0.34 (2H, dd, $J=4.19$ and 4.24 Hz, H-19); ms m/z 532, 517, 472, 412, 288, 175.

OXIDATION OF 1.—Compound **1** (7 mg) was dissolved in CHCl_3 and then treated with Jones reagent (2.5 ml) (prepared by dissolving 5 mg CrO_3 in 1 ml H_2SO_4 and diluting with 2 ml H_2O). The reaction mixture was stirred at room temperature for 24 h, diluted with H_2O , and extracted with CHCl_3 . Removal of solvent yielded **1b**, mp 220–222°; $[\alpha]_D +28.17^\circ$ ($c=0.14$, CHCl_3); ir ν max 3040, 2850, 2750, 1720–1695, and 1380 cm^{-1} ; ^1H nmr (CDCl_3 , 300 MHz) δ 9.71 (1H, s, -CHO), 0.98 (3H, s, Me-29), 0.95 (3H, s, Me-18), 0.90 (6H, d, $J=7.18$ Hz, Me-26 and -27), 0.88 (3H, s, Me-28), 0.86 (3H, d, $J=6.95$ Hz, Me-21), 0.57 and 0.34 (2H, dd, $J=4.21$ and 4.26 Hz, H₂-19); ms m/z 440, 425, 288, 175.

Cycloartan-30-ol [**2**].—Mp 180–182°; $[\alpha]_D +27.38^\circ$ ($c=0.21$, CHCl_3); ir, see text; ^1H nmr (CDCl_3 , 300 MHz) δ 3.62 (2H, ABq, $J=12.18$ Hz, H₂-30), 0.94 (3H, s, Me-18), 0.91 (3H, s, Me-29), 0.90 (6H, d, $J=6.98$ Hz, Me-26 and -27), 0.88 (3H, s, Me-28), 0.86 (3H, d, $J=6.72$ Hz, Me-21), 0.56 and 0.33 (2H, dd, $J=4.21$ and 4.28 Hz, H₂-19); ^{13}C nmr (CDCl_3 , 75.43 MHz) δ 31.91 (C-1), 30.34 (C-2), 29.84 (C-3), 39.80 (C-4), 47.09 (C-5), 21.01 (C-6), 28.10 (C-7), 47.84 (C-8), 20.10 (C-9), 26.21 (C-10), 26.02 (C-11), 35.62 (C-12), 45.13 (C-13), 48.79 (C-14), 32.71 (C-15), 26.52 (C-16), 52.21 (C-17), 17.91 (C-18), 29.81 (C-19), 36.02 (C-20), 18.30 (C-21), 36.42 (C-22), 24.01 (C-23), 39.42 (C-24), 28.21 (C-25), 22.51 (C-26), 22.71 (C-27), 19.33 (C-28), 22.03 (C-29), and 62.98 (C-30); ms m/z 428, 413, 410, 395, 315, 288, 175.

ACETYLATION OF 2.—Compound **2** (5 mg) was refluxed with Ac_2O (2.5 ml) in pyridine (1 ml) for 30 min. The reaction mixture was worked up in the usual manner to yield acetate **2a** (3.5 mg) which was crystallized from $\text{CHCl}_3/\text{MeOH}$, mp 198–200°; $[\alpha]_D +24.2^\circ$ ($c=0.14$, CHCl_3); ir ν max 3045, 1710, 1375, and 1220 cm^{-1} ; ^1H nmr (CDCl_3 , 300 MHz) δ 3.24 (2H, ABq, $J=12.78$ Hz, H₂-30), 2.12 (3H, s, OAc), 0.95 (3H, s, Me-18), 0.92 (3H, s, Me-29), 0.90 (6H, d, $J=6.99$ Hz,

Me-26 and -27), 0.87 (3H, s, Me-28), 0.86 (3H, d, $J=6.81$ Hz, H₂-19); ms m/z 470, 410, 455, 288, 175.

OXIDATION OF 2.—Compound **2** (7 mg) was dissolved in CHCl_3 and then treated with Jones reagent (2.5 ml). The reaction mixture was stirred at room temperature for 24 h, diluted with H_2O , and extracted with CHCl_3 . Removal of solvent yielded **2b**, as colorless needles, mp 204–206°; $[\alpha]_D +21.46^\circ$ ($c=0.17$, CHCl_3); ir ν max 3040, 2850, 2750, 1710, and 1375 cm^{-1} ; ^1H nmr (CDCl_3 , 300 MHz) δ 9.72 (1H, s, -CHO), 0.97 (3H, s, Me-29), 0.95 (3H, s, Me-18), 0.90 (6H, d, $J=7.18$ Hz, Me-26 and -27), 0.88 (3H, s, Me-28), 0.87 (3H, d, $J=7.02$ Hz, Me-21), 0.52 and 0.33 (2H, dd, $J=4.22$ and 4.28 Hz, H₂-19); ms m/z 426, 398, 411, 288, 175.

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